

Determination of Acid number and free fatty acids (FFA) in fats and oils



#### Use

The method is suitable for edible fats and oils such as butter, olive, palm or sunflower oil. The acid number is the quantity of base, expressed in milligrams of potassium hydroxide, that is required to neutralize all acidic constituents present in 1 g of sample. The calculation of the % FFA depends on the titrated type of sample.

### Appliances

- Titrator: TL 7000/TL 7750 M1
- Basic device
- Magnetic stirrer TM 235
- 10 mL Exchange unit WA 10, with amber glass bottle for the titrant, complete

#### Electrodes

- Electrode: N 6480 eth
- Electrolyte: L 5034 (LiCl/ethanol)
- Calibration: n.a.



#### Reagents

- Titrant: KOH 0.1 mol/l in IPA (2-propanol). Also KOH 0.1 mol/l in ethanol
- Titer determination: Potassium hydrogen phthalate
- Solvent: Ethanol/diethyl ether (1:1)

#### Description

#### Determination of the exact concentration of the KOH titrant

We recommend ready to use KOH titrants. The exact concentration of the KOH 0.1 mol/l can be determined using the titrimetric standard potassium hydrogen phthalate.

In a 150 mL beaker, 0.2 g of the standard are weighed accurately and dissolved in 80 mL of dist. water with stirring. It is titrated with the 0.1 mol/l KOH solution.

Repeat the standardization two times. The average value is stored automatically in the exchangeable unit.



Pic. left: titer



Page 1: Curve and result: Titer determination



Calculation formula

Titer: (W\*F2)/((EQ1-B)\*M\*F1) -> M103 Mol (M): 204.22000

Weight (W):	man	Factor 2 (F2):	1000.0000
Blank value (B):	0.0000 ml	Factor 1 (F1):	1.0000
Statistics:	Off		



### Page 2: Method parameters Titer determination:

#### Method data overall view

Method name:	Titer KOH	Created at:	09/19/12 17:05:06	
Method type:	Automatic titration	Last modification:	09/19/12 17:32:02	
Measured value:	mV	Damping settings:	None	
Titration mode:	Dynamic	Documentation:	GLP	
Dynamic:	Steep			
Measuring speed / drift:	Normal:	minimum holding time:	02 s	
		maximum holding time:	15 s	
		Measuring time:	02 s	
		Drift:	20 mV/min	
Initial waiting time:	0 s			
Titration direction:	Decrease			
Pretitration:	Off			
End value:	Off			
EQ:	On (1)			
Slope value:	Steep	Value:	700	

100 %	Filling speed:	30 s
50.00 ml		
10ml		
00072696		
TBA Hydroxid		
1.0265		
0.10320		
09/20/12 0:57:27		
04/12/12		
10/19/11		
12/01/10		
09/21/12 15:13:56		
	100 % 50.00 ml 10ml 00072696 TBA Hydroxid 1.0265 0.10320 09/20/12 0:57:27 04/12/12 10/19/11 12/01/10 09/21/12 15:13:56	100 % Filling speed: 50.00 ml 10ml 00072696 TBA Hydroxid 1.0265 0.10320 09/20/12 0:57:27 04/12/12 10/19/11 12/01/10 09/21/12 15:13:56



#### Titration of the sample

Weigh the sample in a 100 ml beaker and add at least 50 ml of the solvent mixture to the sample. If necessary heat the solution to dissolve the sample.

The sample weight should be calculated and selected that the titration amount is not more than 5 ml because of the long titration time.

For acid numbers between 0.2 and 1 the sample amount should be about 10 - 20 g. For acid numbers between 1 and 10 the sample amount should be about 1 - 3 g.

Place the beaker on the magnetic stirrer and start the titration method. After the titration rinse the electrode and burette tip with solvent. For each set of samples perform a blank titration with 50 ml of the titration solvent.

#### **Result calculation**

The enclosed titration example shows the calculation of the result in mg KOH /g sample (acid number).

The calculation of the % FFA value depends on the titrated sample. For many oil and fat samples the molecular weight of the oleic acid (282 g/mol) is used.

% FFA = (EQ1-B) \* 282 \* T \*100 /(1000\*W)

- EQ1: ml consumption at the equivalence point
- B: ml consumption for the blank titration
- 282: molecular weight of oleic acid in g/mol
- T: concentration of the KOH titrant (e.g.0.1 mol/l)
- 100: per 100 g sample
- 1000: conversation
- W: sample weight in g



### Blank titration page 1: Curve and result



Calculation formula	
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Blank:	EQ1 -> M02	
Statistics:	Off	

Statistics: Off



### Blank titration page 2: method

#### Method data overall view

Method name:	Blank AN	Created at:	04/29/13 16:44:04
Method type:	Automatic titration	Last modification:	04/29/13 16:46:25
Measured value:	mV	Damping settings:	strong
Titration mode:	Linear	Documentation:	GLP
Linear steps:	0.010 ml		

Measuring speed / drift: 12 s

Initial waiting time:	10 s
Titration direction:	Decrease
Pretitration:	Off
End value:	Off
EQ:	Off

Test according ISO 8655: 12/01/10

Last modification: 09/21/12 15:28:02

Dosing	par	ameter	
Dooning	pull	unnocor.	

Dosing speed:	100.00 %	Filling speed:	30 s	
Maximum dosing volume:	0.30 ml			
Unit values				
Unit size:	10ml			
Unit ID:	00072696			
Reagent:	TBA Hydroxid			
Batch ID:	1.0265			
Concentration [mol/l]:	0.10350			
Determined at:	09/21/12 22:27:50			
Expire date:	04/12/12			
Opened/compounded:	10/19/11			



### Sample titration page 1: Curve and result



Calculation formula		
AN mg KOH/g:	(EQ1-B)*T*M*F1/(W*F2)	Mol (M):
Blank value (B):	0.0990 ml (M02)	Titre (T):
Factor 1 (F1):	1.0000	Weight (W):
Factor 2 (F2):	1.0000	Statistics:

56,10000

Off

0.10350000 (a) 10.03650 g (m)



### Sample titration page 2: method

Method data overall view			
Method name: Method type: Measured value: Titration mode: Linear steps:	Acid number Automatic titration mV Linear 0.050 ml	Created at: Last modification: Damping settings: Documentation:	04/29/13 16:20:59 04/29/13 16:46:51 strong GLP
Measuring speed / drift:	User-defined:	minimum holding time: maximum holding time: Measuring time: Drift:	07 s 20 s 04 s
Initial waiting time:	10 s	<b>D</b> iffe	20 1111/1111
Titration direction:	Decrease		
Pretitration:	Off		
End value:	Off		
EQ:	On (1)		
Slope value:	Flat	Value:	120

Dosing parameter			
Dosing speed:	100.00 %	Filling speed:	30 s
Maximum dosing volume:	6.00 ml		
Unit values			
Unit size:	10ml		
Unit ID:	00072696		
Reagent:	TBA Hydroxid		
Batch ID:	1.0265		
Concentration [mol/l]:	0.10350		
Determined at:	09/21/12 22:27:50		
Expire date:	04/12/12		
Opened/compounded:	10/19/11		
Test according ISO 8655:	12/01/10		
Last modification:	09/21/12 15:28:02		



#### Notes

If you have any questions on the application, you can feel free to contact us..

优莱博技术 (北京)有限公司

